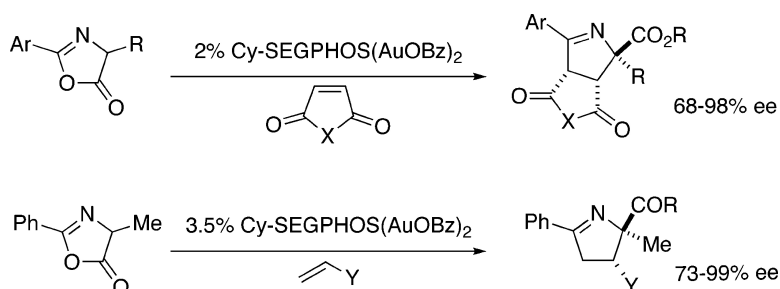


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Au(I)-Catalyzed Enantioselective 1,3-Dipolar Cycloadditions of Münchnones with Electron-Deficient Alkenes

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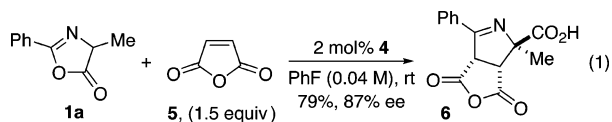
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Synthetic methods relying on gold complexes as catalysts have recently been the focus of intense development.¹ Despite numerous advances, relatively few enantioselective gold-catalyzed transformations have been described. The earliest example of an enantioselective gold(I)-catalyzed transformation, the Hayashi–Ito aldol reaction, was proposed to rely on activation of the nucleophile as a chiral monophosphineAu(I) enolate.² In contrast, the majority of recently reported methods rely on the electrophilic nature of cationic bisphosphinegold(I) complexes to activate π -bonds toward addition of nucleophiles.³ Therefore, the utility of chiral bisphosphinegold(I) complexes would be significantly extended if they could be employed as catalysts for enantioselective transformations that are not predicated on π -bond activation. To this end, herein we describe the development of a bisphosphinegold(I)-catalyzed enantioselective 1,3-dipolar cycloaddition⁴ reaction of mesoionic azomethine ylides (münchnones) with alkenes.

We were inspired by Tepe's recent report of silver(I)acetate-catalyzed münchnone/alkene 1,3-dipolar cycloadditions,^{5,6} to consider the use of our recently developed bisphosphinegold(I) carboxylate complexes as catalysts for this transformation.^{3c} Gratifyingly, treatment of a THF solution of azlactone **1a** and 1.5 equiv of *N*-phenylmaleimide (**3**) with 2 mol % triphenylphosphinegold(I) benzoate at room temperature, followed by in situ esterification, afforded the desired Δ^1 -pyrroline (\pm)-**2a** in 85% yield with excellent diastereoselectivity (Table 1, entry 1).

Having established an achiral phosphinegold(I) benzoate as a catalyst for the formation for **2a**, we next focused on the enantioselective reaction of **1a** and **3** (Table 1). In general, the diphenyl-substituted biarylphosphinegold(I) benzoate complexes successfully employed in the hydroamination gave modest selectivity for the reaction of **1a** and **3** (entries 2–5). While substitution on the phosphine aryl ring resulted in improved selectivity (entry 8), we were pleased to find the (*S*)-Cy-SEGPHOS(AuOBz)₂ (**4**)⁷-catalyzed cycloaddition gave **2a** with a notable increase in enantioselectivity (88% ee) (entry 9). Further optimization of reaction conditions revealed that the reaction performed similarly in various solvents;⁸ however, employing fluorobenzene (PhF) as the solvent produced a further improvement and provided **2a** in 76% yield and 95% ee (entry 10).⁹ Notably, in all cases only the *exo*-adduct was observed.



A variety of electron-deficient alkenes were found to be viable dipolarophiles in the gold(I)-catalyzed münchnone cycloaddition. For example, 2 mol % gold(I) benzoate **4**-catalyzed the reaction of **1a** with maleic anhydride (**5**), under conditions similar to those used with *N*-phenylmaleimide to afford acid **6** in 79% yield and

Table 1. Development of the Au(I)-Catalyzed Cycloaddition^a

entry	catalyst	time (h)	yield (%) ^b	ee (%) ^c
1	PPh ₃ AuOBz	0.5	85	–
2	(<i>R</i>)-BINAP(AuOBz) ₂	2	78	–8
3	(<i>R</i>)-SEGPHOS(AuOBz) ₂	3	81	–40
4	(<i>R</i>)-DIFLUOROPHOS(AuOBz) ₂	3	56	–44
5	(<i>R</i>)-Cl–MeO–BiPHEP(AuOBz) ₂	3	80	–5
6	(<i>R</i>)-SYNPHOS(AuOBz) ₂	3	62	–55
7	(<i>R</i>)-3,5-xylyl-BINAP(AuOBz) ₂	3	63	–24
8	(<i>R</i>)-DTBM-SEGPHOS(AuOBz) ₂	7	81	–83
9	(<i>S</i>)-Cy-SEGPHOS(AuOBz) ₂ , (4)	2	70	88
10	4	5	76	95 ^d

^a Reactions run with 0.33 mmol **1a**; for conditions concerning in situ ester formation see Supporting Information (SI). ^b Isolated yield. ^c Absolute configuration determined by X-ray crystallography on the corresponding α -methylbenzylamide (see SI). ^d Reaction performed at 0.5 M in PhF.

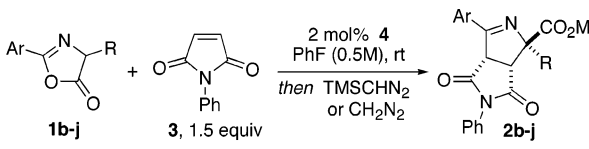
Table 2. Reaction of **1a** with Various Acyclic Alkenes^a

entry	product	time (h)	yield (%) ^b	ee (%)
1	7a , X = CO ₂ ^t Bu R = OMe	24	56	99 ^c
2	7b , X = CO ₂ ^t Bu R = NHCH ₂ Ph	14	74	95
3	7c , X = CO ₂ Et R = OMe	14	66	90
4	7d , X = CO ₂ Me R = OMe	14	89 ^d	93
5	7e , X = CN R = NHCH ₂ Ph	14	68	76

^a Reactions run with 0.33 mmol **1a** at 0.5 M; for conditions concerning in situ ester/amide formation see SI. ^b Isolated yield unless otherwise noted. ^c Reaction run with 10 equiv of alkene and 2% mol **4**. ^d Yield determined by ¹H NMR.

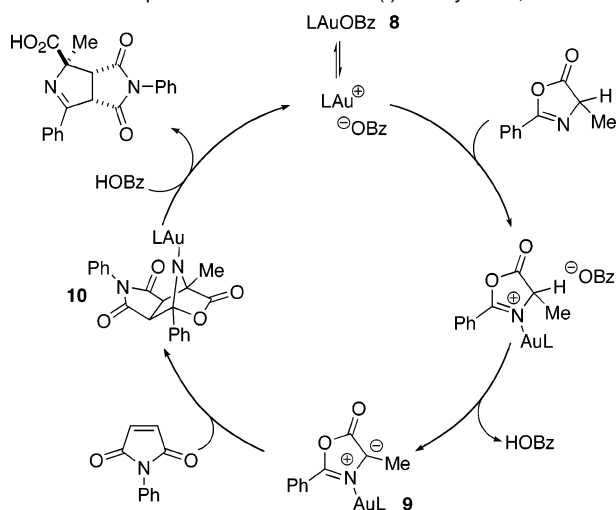
87% ee (eq 1).¹⁰ Acyclic alkenes could also be employed as partners in the gold(I)-catalyzed cycloaddition (Table 2). Generally, the reactions performed best using a 3:1 mixture of THF and PhF and slightly higher catalyst loading (3.5%). Notably, in all cases the reactions proceeded with excellent diastereo- and regioselectivity.

Various azlactones (**1b–j**) were prepared, and the results of their gold(I) benzoate-catalyzed enantioselective cycloaddition with *N*-phenylmaleimide are shown in Table 3. Substitution at the para position of the azlactone aromatic ring was well tolerated (entries 1–4). While increasing the steric demand at the azlactone C2 or C4 position resulted in decreased reactivity in PhF, switching the solvent to a mixture of THF and PhF permitted isolation of the cycloadducts in good yields (entries 5, and 7–9). The use of this solvent mixture allowed for smooth reaction of azlactone **1f**, bearing

Table 3. Reaction of *N*-Phenylmaleimide (**3**) with Azlactones^a


entry	product	time (h)	yield (%) ^b	ee (%)
1	b , R = Me, Ar = <i>p</i> -MeO-C ₆ H ₄ -	18	77	95
2	c , R = Me, Ar = <i>p</i> -Br-C ₆ H ₄ -	15	75	93
3	d , R = Me, Ar = <i>p</i> -Cl-C ₆ H ₄ -	15	72	92
4	e , R = Me, Ar = <i>p</i> -NO ₂ C ₆ H ₄ -	1.5	98	91
5	f , R = Me, Ar = <i>o</i> -Me-C ₆ H ₄ -	4	73	86 ^c
6	g , R = H, Ar = Ph	24	84	-98 ^d
7	h , R = allyl, Ar = Ph	8	86	87 ^c
8	i , R = Ph, Ar = Ph	1.5	35	78 ^e
9	j , R = Bn, Ar = Ph	36	71	68 ^c

^a Reactions run with 0.33 mmol azlactone; for conditions concerning in situ ester formation see SI. ^b Isolated yield. ^c Run at 0.5 M in 3:1 THF/PhF. ^d At 0.5 M in acetone, using (*R*)-DTBM-SEGPHOS(AuOBz)₂; note also the change in the sense of ligand chirality. ^e Run at 0.25 M in 3:1 THF/PhF at 0 °C with 5% mol **4**.

Scheme 1. Proposed Mechanism of Au(I)-Catalyzed 1,3-DCR

a 2-methylphenyl C2 substituent, providing product **2f** in good yield with 86% ee.¹¹ Similarly, C4 allyl-substituted azlactone **1h** underwent gold(I)-catalyzed cycloaddition to furnish **2h** in 86% yield and 87% ee; however, a decrease in enantioselectivity was observed with a further increase in the size of the C4 substituent to benzyl (entry 9). The reaction of **3** with glycine-derived azlactone **1g** catalyzed by **4** produced **2g** with only 81% ee. Fortunately, switching the catalyst to (*R*)-DTBM-SEGPHOS(AuOBz)₂ allowed for the isolation of cycloadduct **2g** in 84% yield and 98% ee (entry 6).

A proposed catalytic mechanism, paralleling those postulated for reactions of acyclic azomethine ylides, is shown in Scheme 1.⁶ Dissociation of a carboxylate counterion from **8** provides an open coordination site for azlactone binding. Deprotonation of the activated substrate, presumably by benzoate, generates *N*-aurated-dipole **9**. Reaction of **9** with the dipolarophile produces initial cycloadduct **10**. Subsequent C–O bond cleavage and protonation followed by dissociation of the Δ¹-pyrroline regenerates the catalyst.¹²

In summary, we have developed the first catalytic enantioselective reaction of azlactones with alkenes to provide Δ¹-pyrrolines.^{13,14} Notably, the gold-catalyzed cycloadditions proceed with excellent diastereo- and regioselectivity. The reaction is proposed to proceed through a 1,3-dipole¹⁵ generated by deprotonation of a gold(I)-

activated azlactone and therefore represents an important departure from the mechanistic paradigm of π-activation most commonly proposed in contemporary asymmetric catalysis with gold complexes. The development of enantioselective reactions relying on gold(I)-catalyzed generation of nucleophiles is ongoing and will be reported in due course.

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Supporting Information Available: Experimental procedures and compound characterization data; X-ray crystallographic data. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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- (8) THF (2 h, 70%, 88% ee), acetone (2 h, 82%, 85% ee), DME (3 h, 79%, 87% ee), CH₂Cl₂ (5 h, 59%, 88% ee), benzene (5 h, 64%, 88% ee), toluene (5 h, 62%, 87% ee).
- (9) Under these conditions, the choice of carboxylate counterion had no notable effect on the reaction. (benzoate, 5 h, 76%, 95% ee; *p*-nitrobenzoate, 5 h, 70%, 95% ee; acetate, 5 h, 71%, 95% ee).
- (10) Cycloadduct **6** was not isolated; yield determined by ¹H NMR (see SI).
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